EFFECT OF N FERTILIZER FORMS AND SOIL MOISTURE LEVELS ON THE N GASEOUS LOSSES

*NÓTÁS, E.¹ – MOLNÁR, E.¹ – RUZSA, D.¹ – CSOMA, Z.² – DEBRECZENI, K.³ – HELTAI, GY.¹

¹ Szent István University, Faculty of Agricultural and Environmental Sciences, Department of Chemistry and Biochemistry, H-2103 Gödöllő, Páter Károly u. 1. (phone: +36-28-522-000/1668; fax: +36-28-410-804)

²Ferenc Rákóczi II. Transcarpathian Hungarian Institute, Department of Biology and Chemistry, 90202 Beregovo, Kossuth tér 6, Ukraine

³Pannonia University, Georgikon Faculty of Agricultural Sciences, H-8360 Keszthely, Deák F. u. 16.

> *Corresponding author e-mail: notas.erika@mkk.szie.hu

(Received 22nd Apr 2014; accepted 22nd July 2014)

Abstract. In a pot experiment using brown forest soil with clay eluviation (40 kg soil/pot) from Keszthely the gaseous losses of nitrogen (total N, N₂, N₂O NO NO₂) caused by denitrification were tested. Gas-collecting traps were placed at a depth of 20 cm in the soil. At the same fertilizer N input (6250mgN/pot, 150mg N/kg soil), the effects of two fertilizer forms (KNO₃, NH₄Cl) and two soil moisture levels (field water holding capacity, WHC= 65% and 100%) were evaluated to the gaseous losses of nitrogen with and without sowing of maize hybrids (two Stira and two My 355/pot) as test plants. The composition of the N-containing gases was determined by gas chromatography. From the soilatmosphere the gas samples were taken 8 times, the sampling times were on the 2, 16, 30, 54, 68, 82, 110, 138-th days of the experiment. Statistical evaluation of the results was performed by analysis of variance. The N gaseous loss in the planted pots was the 12% of the applied fertilizer dose, while in unplanted pots it was nearly the double of it (22%) Consequently, there was a 50% decrease in the denitrification gaseous losses of fertilizer N due to plant N uptake. The N-gaseous loss was significantly higher in cases of KNO₃ applications and at the WHC=100% soil moisture than in NH₄Cl treatments and at the WHC= 65%, probably due to more intensive denitrification. In the unplanted pots the amount of N_2 and N_2O increased the most (by 95% and by 70%) compared with planted pots. It can be concluded that the improperly applied fertilizer causes significant GHG emissions in the absence of plants. Keywords: N-gaseous losses, nitrous-oxide, GHG emission, fertilizer treatment, soil moisture

Introduction

The relative contributions of the three main greenhouse gases (GHG): CO₂, CH₄ and N₂O to overall global radiative forcing are 63.5%, 18.1% and 6.2%, respectively. Soils are one of the major sources of GHG contribution both to anthropogenic emissions due to land use change and agricultural management (up to 22.5% of all anthropogenic sources) (Blagodatsky and Smith, 2012). Yienger and Levy (1995) estimate for annual above-canopy emissions is 5.5Tg N (NO_x) with a range of 3.3-7.7 Tg N. Globally the strongest emitters are agriculture, grasslands, and tropical rain forests, accounting for 41%, 35% and 16% of the annual budget, respectively."Pulsing" (the emissions burst following the wetting of a dry soil) contributes 1,3 Tg N annually. In temperate regions, agriculture dominates emission, and in tropical regions, grassland dominates. By the

year 2025, the increasing use of nitrogen fertilizer may raise total annual emissions to 6.9 Tg N with agriculture accounting for more than 50% of the global source. The agriculture is a significant contributor to the increasing N₂O concentration of air (Delmas et al., 1997; Hellebrand et al., 2003; Ruser et al., 2006). The agriculture is estimated to contribute more than two-thirds of total anthropogenic N₂O emissions (Pattey et al., 2007). The agricultural soils emit 3.3 Mt N₂O per year (Mosier et al., 1998). As the global warming potential of N_2O is 298 times as strong as that of CO_2 , N_2O emissions from soils thereby exert an important impact on the environment. Soil microbial processes, nitrification and denitrification contribute significantly to the agricultural N₂O emission (Stange and Döhling, 2005). N-gaseous emissions from agricultural soils, speciation of predicted gas N flux into N₂O and N₂ depend on the Nfertilizer dose and type, the soil type and pH, the soil moisture, the oxygen supply and the C/N ratio of the soil, application practice, the vegetation and the temperature (Ma et al., 2007; Chu et al., 2007, Gu et al, 2009). Field response to fertilizaton is variable; some plots have enhanced emissions for prolonged periods, whereas others have sharper initial increases that decay over time. In general, however a positive linear correlation seems to be between fertilizer use and emission, and over the course of a growing season, total NO_x emissions change between 1-10% of added nitrogen fertilizer (Williams et al., 1992; Shepard et al., 1991, Cardenas et al., 1993). The exact fertilizer form which is more stimulatory depends on the site, although evidence suggests that a mixed form, such as NH₄NO₃, generates strongest emission (Sanhueza, 1992). In a pot experiment the total N gaseous losses were higher at KNO₃ treatment than NH₄Cl treatment in brown forest soil with clay eluviation (Debreczeni et al. 1998). The N₂O emissions from NH4⁺-N fertilizer were higher than from urea and NO3⁻-N fertilizers in coarse- textured soils (Gu et al. 2009). Kramer et al.(1999) studied greenhouse gas emissions of Dutch agricultural crop production by using a life cycle approach. It was determined that the emissions of N₂O are mainly caused by the production and application of synthetic nitrogen fertilizer. The use of other nitrogen sources manure, compost or limiting the use of synthetic nitrogen could positively influence the emissions of N₂O. In contrast, Jones et al. (2005) found that during growing season the cumulative total N₂O flux from manure treatments was 25 times larger than that from mineral fertilizers. Large inputs of manure N have a potential to contribute a very high N_2O emissions, particularly in wet soil conditions. The ratio of NO_x gases depends on the soil moisture. In an experiment testing the tropical rain forest ecosystem during dry periods the NO-N: N_2O-N ratio was as high as 60:1, whereas for wetter periods it decreased to <7:1. (Butterbach et al, 2004). In Hungary, where the proportion of agricultural area to the total area of the country is quite high (63%) it is essential to investigate agricultural contribution to greenhouse gas emission.

Material and methods

Brown forest soil with clay eluviation from Keszthely (pH_{KCl} =7.7, C=1.1 %, NO₃-N = 17.5 mg kg⁻¹, NH₄-N = 12 mg kg⁻¹, humus = 1.9 %) was filled in large plastic pots (40 kg soil/pot). N fertilizers (KNO₃, NH₄Cl) were applied at a high rate of approx. 150 mg N kg⁻¹ soil (6 250 mg N/pot) to trace the precise transformation processes of N, P fertilizer in all treated soils, and K fertilizer in NH₄Cl treatments was also used (N:P:K= 1:1:3.35). All experiments were carried out at two soil moisture levels (field water holding capacity, WHC= 65 % and 100 %) with and without sowing of maize as test

plants (two Stira and two Mv355 plants/pot) (Table1). For statistical calculations, results of maize hybrids were taken as replicates. The plants were grown until full ripening. Maize (*Zea mays L.*) was sown in 48 pots, and 32 pots were left without plant.

	With maize					Without maize							
Maize hybrid	Treatment	Control		KNO ₃ +P		NH ₄ Cl+PK		Control		KNO ₃ +P		NH ₄ Cl+PK	
Stira	WHC %	100	65	100	65	100	65	100	65	100	65	100	65
MV 355	WHC %	100	03	100	03	100	03	100	03	100	03	100	03

Table 1. Setting up of the experiment

All treatments were replicated four times. Gas-collecting traps of 1800 cm³ capacity with silicon pipe outlets were placed at a depth of 20 cm in the soil. The traps collected the N gaseous losses (NO, NO₂, N₂O, N₂) released during the denitrification process. Gas samples were taken 8 times by syringe (5 cm³) in 3 replicates from the soil atmosphere. For soil atmosphere analysis the composition of the N-containing gases (N₂, NO, NO₂, N₂O) was determined by a Carlo Erba 2350 type gas chromatograph. From these data the total gaseous N losses were calculated (mg N /pot or mg N/1800 cm³) for each sampling point. At the end of the experiments the cumulative gaseous N losses and N₂O-N losses were also calculated by the difference method (N content difference between treated and control gas samples) and by the calculations of the relative % gaseous N losses. Results were analyzed by the means of analysis of variance (MANOVA). The statistical analyses were performed with a Microsoft Excel Macro (Aydinalp et al., 2008; Vágó et al., 2008)

Results and discussion

The total N and total N₂O gaseous losses were found to depend significantly $(LSD_{5\%}=16,7 \text{ and } 5,1)$ on the presence or absence of the plants (*Table 3*). In control samples in the planted pots the total N gaseous losses were 1,4 times higher (at 65% and 100% WHC), the N₂O gaseous losses were 1,6 times (at 65%WHC) and 1,9 times (at 100 %WHC) higher than in the unplanted pots (*Table 2*). Similar results were obtained by Heltai et al. (2013) in a pot experiment in control samples. In contrast to the control, in the unplanted pots treated with fertilizer the total N and N₂O gaseous losses were higher (1,4-1,7 times) than in the planted pots. Calculated by difference method the total N gaseous losses were higher in the unplanted pots by 78-84% at KNO₃ treatment and by 71-73% at NH₄Cl treatment while the total N₂O gaseous losses were higher by 67-72% and by 62-63% compared with planted pots.

The total N and total N₂O gaseous losses were significantly higher in the N-fertilizer treatments than in control samples (*Table 3*: LSD_{5%}=20,5 and 6,3). The difference of total N gaseous losses was 7 times higher in KNO₃ and NH₄Cl treatment in the planted pots while in the unplanted pots it was 16-17 times higher than in the controls (*Table 2*). The difference of N₂O gaseous losses was 5-6 times higher in the planted pots and 14-15 times higher in the unplanted pots in fertilized pots than in the controls. The total N and total N₂O gaseous losses were more intensive in case of KNO₃ treatment than in NH₄Cl treatment (*Table 3*: LSD_{5%}=20,5 and 6,3).

Treatment	Maize	WHC %	∑N	$\sum N_2 O$
Control	Ŧ	100	131,5	20
Control	Т	65	119,5	16,5
KNO	Ŧ	100	921	107
KINU3	т	65	868,5	96
	1	100	907,5	102
NH4CI	Ŧ	65	868,5	93,5
Control		100	91,5	10,5
Control	-	65	85,5	10
LNO		100	1494	156
KINU ₃	-	65	1461,5	146,5
		100	1438	144,5
NH4CI	-	65	1369	135

Table 2. Total N and total N_2O gaseous losses (Nmg/1800cm³)(+ with plant, - without plant)

Table 3. Analysis of variance of total N and total N_2O gaseous losses according to plant presence, soil moisture and treatment (N and N_2O mg/1800cm³)

Plant presence	ΣN	∑N₂O	Soil moisture (WHC%)	ΣN	∑N₂O	Treatment	ΣN	∑N₂O
Maize +	636	73	100%	831	90	Control	107	14,3
Maizo -	000	100	65%	705	92	KNO ₃	1186	126
IVIAIZE -	990	0 100 05%		795	65	NH₄CI	1146	119
LSD _{5%}	16,7	5,1	LSD _{5%}	16,7	5,1	LSD _{5%}	20,5	6,3

The difference was significant in the unplanted pots only (*Table 4*: $LSD_{5\%}=29$ and 8,8). Similar results were obtained in other pot experiments using same soil samples from Keszthely by Debreczeni et al.(1995, 1998) and Nótás et al. (2003).

Table 4. Analysis of variance of total N and total N_2O gaseous losses according to plant presence x treatment, plant presence x soil moisture and treatment x soil moisture (N and N_2O mg/1800cm³)

Plant presence X Treatment		ΣN	Σ N ₂ Ο	Plant p X Soil r	Plant presence X Soil moisture		∑N₂O	Treatr Soil mo	nent X bisture	ΣN	Σ N ₂ O
Maize	Cont- rol	126	18,3	Maize	100%	653	76,3	Cont- rol	100%	111,5	15
+	KNO₃	895	102	+	65%	619	68,7		65%	102,5	13
	NH ₄ Cl	888	98						100%	1207	132
Maize -	Cont- rol	89	10,3	Maize	100%	1008	103,7	KNO₃	65%	1165	121
	KNO ₃	1478	151	-	65%	972	07.2		100%	1173	123
	NH ₄ Cl	1404	140				57,2		65%	1119	114
LSD _{5%}		29	8,8	LS	LSD _{5%}		7,2	LSD _{5%}		29	8,8

The total N and total N₂O gaseous losses were higher at 100%WHC than 65%WHC (*Table 3*: LSD_{5%}=16,7 and 5,1). This difference of total N loss was equally significant in the planted and unplanted pots (*Table 4*: LSD_{5%}= 23,6), but the total N₂O gaseous loss was significantly higher only in the planted pots (*Table 4*: LSD_{5%}= 7,2). The effect of the soil moisture was significant in the fertilized samples only, in the control samples it was not (*Table 4*: LSD_{5%}=29 and 8,8). The differences were dependant on the presence or absence of the plants in control samples 7-10% and 5-20%, these are not significant; while in KNO₃ treatment 2-6% and 7-12%, in NH₄Cl treatment 5% and 7-9%, these are significant.

Figure 1 and 2 (on the basis of *Table 5*) show the rate of total N gaseous losses increase in the planted pots (*Fig. 1*) and in the unplanted pots (*Fig. 2*). Results of maize hybrids were averaged.

Treatment	Maize hybrid	WHC %		Sampling time (day)								
			2.	16.	30.	54.	68.	82.	110.	138.		
Control	Stira	100	47	70	96	126	127	130	123	122		
		65	37	62	7	108	113	122	112	115		
Control	My 355	100	48	71	91	126	124	133	121	116		
	ININ 222	65	31	65	80	101	105	117	100	109		
KNO3	Stira	100	150	344	480	616	719	931	918	876		
		65	110	305	448	587	619	848	836	844		
	Mv 355	100	124	365	504	615	743	911	903	863		
		65	116	337	474	589	692	889	858	874		
	Stira	100	121	337	475	589	604	918	894	855		
NH CI		65	107	297	426	546	587	846	827	844		
111401	Mv 355	100	119	333	492	595	701	893	897	870		
		65	102	303	475	579	624	850	839	891		
Control		100	40	47	58	76	84	94	86	89		
Control	-	65	38	40	52	78	75	89	73	82		
KNO ₃		100	108	345	651	911	971	1156	1494	1493		
	-	65	101	304	614	829	895	1125	1343	1473		
NH CI		100	108	342	658	891	956	1114	1415	1461		
NH4CI	-	65	93	300	614	824	852	1096	1352	1386		

Table 5. Total N gaseous losses in the sampling time, $\sum Nmg/1800cm3$ (- unplanted pots)

It can be established that the sampling time significantly influences the production rates of total N gaseous losses. The change of total N gaseous losses growth can be characterized by two peaks, in the planted pots (*Fig.1*) on the 16th and 82nd day of the experiment and in the unplanted pots (*Fig.2*) later, on the 30th and 110th day. This difference was probably caused by the increase of microbiological activity due to the effect of root secretion in the planted pots. Similar results were obtained in other pot experiments using the same soil type from Keszthely Debreczeni et al.(1995). The minimum values were received on the 68^{th} day in the planted and unplanted pots equally. The N gaseous emission fluctuations may be caused by periodic changes of soil bacteria's microbiological activity, life cycle and temperature.

In our experiment with maize the total N gaseous losses changed between 749-789 mgN/1800 cm³ in the planted pots and between 1304-1408 mgN/1800 cm³ in the unplanted pots (*Fig. 3*). The total N gaseous losses in the planted pots were 12-13% of

the applied fertilizer N while in the pots without maize it was about twice as much, 21-22%.



Figure 1. The change of total N gaseous losses increase in the planted pots in the function of sampling time



Figure 2. The change of total N gaseous losses increase in the unplanted pots

The change of the relative total N gaseous losses (*Fig.4*) was found to be highly influenced by the presence of maize, the fertilizer treatment and soil moisture. It was higher at KNO₃ treatment and 100% WHC soil moisture (except 30^{th} day), than NH₄Cl treatment and at 65% WHC. Between the 2^{nd} and 68^{th} day the 100%WHC increased more intensively the total N gaseous losses in unplanted pots than with maize compared to 65%WHC while KNO₃ treatment enhanced the gaseous N losses stronger in planted pots than without maize compared to NH₄Cl. The forms of the curves are similar among the 2^{nd} and 68^{th} sampling times. The effect of KNO₃ and 100%WHC is the strongest at

the start of the experiment $(2^{nd} \text{ and } 16^{th} \text{ day})$ and on the 68^{th} day compare to NH₄Cl and 65%WHC.



Figure 3. The total N gaseous losses (+ planted pots, -unplanted pots)(LSD5%= 41mgN/1800 cm3)



Figure 4. Distribution of the relative total N gaseous losses in the planted and unplanted pots

The ratio of NO_x gaseous losses was different in the planted and unplanted pots (*Fig.5*). In the unplanted pots the ratio of N₂ gas was higher (78%) while the ratios of N₂O, NO₂ and NO were lower than in planted pots. The N₂ and N₂O gaseous losses were higher by 95% and by 70% in the unplanted pots than in the planted pots (N₂ and N₂O: LSD_{5%}= 17 and 5,1 N and N₂O mg/1800cm³) (*Fig.6*).



Figure 5. The % distribution of N gaseous losses in planted (left) and unplanted pots (right)



Figure 6. The distribution of N gaseous losses in the planted and unplanted pots

After the 16th day of the experiment the N₂O gaseous losses were much higher in the unplanted pots than in the planted pots and reached the maximum value (140 mgN/1800 cm³) on the 82nd and 110th day of the experiment depending on treatment (*Fig.7*) and soil moisture (*Fig.8*) while in the planted pots the maximum value (about 80 mgN/1800 cm³) was measured on the 82nd day. The N₂O gaseous losses were significantly higher at KNO₃ treatment than NH₄Cl treatment in the unplanted pots between the 54th -110th day while in the planted pots between the 2nd – 30th day and on the 82nd day (*Fig.7*). It was found that at WHC=100% soil moisture the N₂O gaseous losses were significantly higher than at WHC= 65% in all pots in most of the sampling time (*Fig.8*). There are two exceptions: in the unplanted pots on the 54th day when the N₂O gaseous loss was significantly higher at WHC= 65% than at WHC=100% soil moisture.



Figure 7. Change of N2O gaseous losses in the sampling time depend on N fertilizer forms (+in planted pots, - in unplanted pots)



Figure 8. Change of N2O gaseous losses in the sampling time depend on soil moisture (+in planted pots, - in unplanted pots)

Conclusions

The total N and N₂O gaseous losses were significantly higher in the unplanted pots than in the planted pots in treated samples. In our experiment with maize the total N gaseous loss was 12-13% of the applied fertilizer N while in the pots without maize it was about twice as much 21-22%. Plant N uptake reduced the N gaseous losses by about 50%.

In contrast to the treated samples in control samples in the planted pots the total N and N_2O gaseous losses were higher than in the unplanted pots due to soil drying-rewetting intensifying microbiological activity, which is more intensive in the rhizosphere around roots (Kuzyakov et al.2000).

The total N and N_2O gaseous losses were significantly higher in the N-fertilizer treatments than in control samples due to the increased activity of bacteria, accelerated

mineralization and release of nutrients by the large fertilizer N dose (Kuzyakov et al.2000).

The total N and N_2O gaseous losses were more intensive in case of KNO_3 treatment than in NH_4Cl treatment in most cases as an assumed consequence of the intensive denitrification.

The total N and N_2O gaseous losses were significantly higher at 100%WHC than 65%WHC due to more intensive denitrification processes.

In the unplanted pots the ratio of N_2 gas was higher (78%) than in the planted pots (72%). The N_2 and N_2O gaseous losses were higher by 95% and by 70% in the unplanted pots than in the planted pots due to the absence of plants and more intensive denitrification processes.

It can be concluded that the improperly applied fertilizer causes significant GHG emissions in the absence of plants.

The gaseous N losses were found to be influenced significantly by the presence or absence of plants, the N fertilizer form and the soil moisture.

Acknowledgements. The research was supported by Research Centre of Excellence- 17586-4/2013/TUDPOL and TÁMOP-4.2.2.A-11/1/KONV-2012-0007 projects.

REFERENCES

- [1] Aydinalp, C., Füleky, Gy., Tolner, L. (2010): The comparison study of some selected heavy metals in the irrigated and non-irrigated agricultural soils. Bulgarian Journal of Agricultural Science 16: 754-768.
- [2] Blagodatsky S., Smith P. (2012): Soil physics meets soil biology: Towards better mechanistic prediction of greenhouse gas emissions from soil. – Soil Biology and Biochemistry 47: 78-92.
- [3] Butterbach-Bahl K, Kock M, Willibald G, Hewett B, Buhagiar S, Papen H, Kiese R. (2004): Temporal variations of fluxes of NO, NO₂, N₂O, CO₂ and CH₄ in a tropical rain forest ecosystem. – Global Biogeochemical Cycles. 18(3): Article Number GB3012
- [4] Cardenas, L., A. Rondon, Johansson C., Sanhueza E. (1993): Effects of soil moisture, temperature, and inorganic nitrogen on nitric oxide emissions from acidic tropival savannah soils. – J. Geophys. Res., 98(14): 783-790.
- [5] Chu, H., Hosen, Y., Yagi, K. (2007): NO, N₂O, CH₄ and CO₂ fluxes in winter barley field of Japanse Andisol as affected by N fertilizer management. – Soil Biology and Biochemistry 39: 330-339
- [6] Conrad, R. (1996): Soil microoganisms as controllers of atmospheric trace gases (H₂, CH₄, OCS, N₂O, and NO). Micro-biological Reviews 60: 609-640.
- [7] Debreczeni B-né (1995): A nitrogén-műtrágyázás hatása a talajlevegő nitrogéngázösszetételére. – Agrokémia és Talajtan 44(3-4): 299-306.
- [8] Debreczeni K., Fischl K., Heltai GY., Bálint Á. (1998): A nitrogén műtrágyázás hatása a talajból származó különböző nitrogéntartalmú gázokra. – Növénytermelés 47(2): 155-164.
- [9] Delmas, R., Serca, D. Jambert, C. (1997): Global invertory of NO_x sources. Nutrient Cycling in Agroecosystems 48: 51-60.
- [10] Frolking, S. E., A. R., Moiser, D. S., Ojima, C. Li, W. J., Patron, C. S., Potter, E., Piesack, R., Stenger, C. Haberbosch, P., Dörsch, H., Flessa, K., Smith, A. (1998): Comparison of N₂O emissions from soils at three temperate agricultural sites: simulations of year-round measurements by four models. – Nutr. Cycl. Agroecosyst. 52: 77-105.

- [11] Gu, C., Maggi, F., Riley, W.J., Hornberger, G. M., Xu, T., Oldenburg, C.M., Spycher, N., Miller, N. L., Venterea, R. T., Steefel, C. (2009): Aqueous and gaseous nitrogen losses induced by fertilizer application. – Journal of Geophysical Research-Biogeosciences 114: Article Number: G01006
- [12] Hellebrand, H. J., Kern, J., Scholz, V. (2003): Long-term studies on greenhouse gas fluxes during cultivation of energy crops on sandy soils. – Atmospheric Environment 37:,1635-1644.
- [13] Heltai, Gy., Anton, A., Hoffmann, S., Szili-Kovács, T., Berecz, K., Kampfl, Gy., Kristóf, K., Molnár, E., Horváth, M., Bálint, Á. (2013): Ásványi- és szervestrágyázás hatása a CO₂ és N₂O gázok képződésére a talajban. Agrokémia és Talajtan 62(1): 143-162.
- [14] Jones, S. K., Rees, R. M., Skiba, U. M., Ball, B. C. (2005): Greenhouse gas emissions from a managed grassland. – Global and Planetary Change 47: 201-211.
- [15] Kramer, K. J., Moll, H. C., Nonhebel, S. (1999): Total greenhouse gas emissions related to the Dutch crop production system. – Agriculture, Ecosystems and Environment 72: 9-16.
- [16] Kuzyakov, Y., Friedel, J. K., Stahr, K. (2000): Review of mechanisms and quantification of priming effects. – Soil Biology and Biochemistry 32: 1485-1498.
- [17] Ma, W. K., Schautz, A., Fishback, L-A. E., Bedard-Haughn, A., Farrell, R. E., Siciliano, S.,D. (2007): Assessing the potential of ammonia oxidizing bayteria to produce nitrous oxide in soils of a high arctic lowland ecosystem on Devon Island, Canada. – Soil Biology and Biochemisty 39: 2001-2013.
- [18] Mosier A. R., Duxbury J. M., Freney J. R., Heinemeyer O., Minami K. (1998): Assessing and mitigating N₂O emissions from agricultural soils. Climatic Change 40: 7-38.
- [19] Nótás, E., Debreczeni, K., Fischl ,K., Heltai, GY. (2003).: Különböző nitrogén műtrágyák és eltérő talajnedvesség szintek hatása a talaj/növény/légkör rendszer N- mérlegére. – Növénytermelés 52(6): 667-678.
- [20] Pattey, E., Edwards, G. C., Desjardins, R. L., Pennock, D. J., Smith, W., Grant, B., MacPherson, J. I. (2007): Tools for quantifying N₂O emissions from agroecosystems. – Agric. For. Meteorol. 142: 103-119.
- [21] Ruser, R., Flessa, H., Russow, R., Schmidt, G., Buegger, F., Munch, J. C.,(2006): Emission of N₂O, N₂ and CO₂ from soil fertilized with nitrate: effect of compaction, soil moisture and rewetting. – Soil Biology and Biochemistry 38: 263-274.
- [22] Sanhueza, E.,(1992): Biogenic emissions of NO and N2O from tropical savanna soils, in Proceedings of International Symposium on Global Climate Change, pp. 22-34.
- [23] Shepard, M. F., Barzetti, S., Hastie, D. R. (1991): The production of NO_x and N₂O from a fertilized agricultural soil. Atmos Environ., 25: 1961-1969.
- [24] Stange, F., Döhling, F. (2005): ¹⁵N tracing modell SimKIM to analyse the NO and N₂O poduction during autotrophic and heterotrophic nitrification in soils. Isotopes in Environmental and Health Studies. 41(3): 261-274.
- [25] Vágó, I., Tolner, L., Eichler-Löbermann, B., Czinkota I., Kovács, B. (2008): Longterm effects of liming on the dry matter production and chemical composition of
- perennial ryegrass (Lolium perenne L.). Cereal Research Communications. 36, 103-106.
- [26] Williams, E. J., Guenther, A., Fehsenfeld, F. C. (1992): An inventory of nitric oxide emissions from soils in the United States. J. Geophys. Res., 97:7511-7519.
- [27] Yienger, J. J., Levy, H. (1995): Empirical model of global soil-biogenic NO_x emission. Journal of Geophysical Research. 100: NO.D6, 11,447-11,464