SYNTHESIS, CHARACTERIZATION AND INVESTIGATION OF ANTIMICROBIAL ACTIVITY OF SILVER NANOPARTICLES FROM CYDONIA OBLONGA LEAF

BARAN, M. F.

Medical Laboratory Techniques, Vocational Higher School of Healthcare Studies, Mardin Artuklu University, 47200 Mardin, Turkey (e-mail: mfiratbaran@gmail.com)

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Abstract. In this study, leaf extract of *Cydonia oblonga* were used to synthesize silver nanoparticles. Silver nanoparticles have been successfully synthesized by environmentally friendly and economical methods. Characterization of synthesized AgNPs used UV-visible spectroscopy (UV-Vis.), thermogravimetric and differential thermal analysis (TGA-DTA), X-ray diffraction spectroscopy (XRD), scanning electron microscopy (SEM), electron diffraction X-ray (EDX), Fourier transform infrared spectroscopy (FT- IR). XRD results demonstrated that the crystal structure of AgNPs was cubic and had a size of 27.30 nm. AgNPs were found to be spherical in SEM images. The element composition was also revealed by EDX. Antimicrobial activity of silver nanoparticles; Gram-negative *Escherichia coli* was tested on gram positive *Staphylococcus aureus* and *Candida albicans* yeast. The concentration values of 0.0552, 0.1535 and 0.0383 mg L⁻¹ were determined as MIC (minimum inhibitory concentration), respectively. It was evaluated that the commercial antibiotics used were more effective at low concentrations compared to colistin, vancomycin and fluconazole.

Keywords: XRD, SEM, TGA-DTA, silver nanoparticles, Cydonia oblonga

Introduction

Particles with a size of 1-100 nm are called nanoparticles. These particles can be obtained by biological, physical and chemical means. They are used in different areas such as energy, pharmacology, biomedical, cosmetics, textiles, food and agriculture etc. (Chaudhry et al., 2018). Compared to physical and chemical methods, the advantages of biological methods such as ease of application and control, eco- friendliness and lack of toxic chemicals make this method important (Pantidos and Horsfall, 2014). Due to the fact that it is cheap and easy to obtain, the use of vegetable resources for synthesis is becoming increasingly common. In the synthesis of silver nanoparticles, phytochemicals in plant leaf extracts form AgNPs by reducing the Ag⁺ ions in the media to the Ag^o form, and these phytochemicals also provide stability (Prakash et al., 2013). Silver nanoparticles (AgNPs) were synthesized using plant extracts and these particles showed good antimicrobial activity (Ahmed et al., 2018).

In this study, AgNPs were obtained in a very short time with stable, energy-free and room conditions with *Cydonia oblonga* leaf extract and without any additional chemicals/and or physical steps. The resulting nanoparticles showed a strong antimicrobial effect. In this study a simple and rapid green synthesis of AgNPs is presented.

AuNPs using leaves extract of R. rugosa have been reported. Silver and gold nanoparticles can be prepared with lower amounts of leaf extract and without any additional chemicals/and or physical steps. The effect of leaf extract quantity and concentration of metal solution were also evaluated to optimize the synthesis route producing the metal nanoparticles.

Materials and methods

Silver nitrate (AgNO₃) 99.98% was obtained from Sigma-Aldrich. Vancomycin, fluconazole and colistin were commercially purchased for antimicrobial applications.

Preparing Cydonia oblonga leaf extract

The leaf extract of *Cydonia oblonga* was washed with distilled water and dried at room temperature. 25 g was taken and the size reduced. It was mixed with 500 ml distilled water and boiled at 85 °C. The extract was filtered through Whatmann No.1 filter paper and cooled to room temperature. Stored at 4 °C for use in the study.

Synthesis of AgNPs

1 mM AgNO₃ aqueous solution was used for silver nanoparticles synthesis. It was left to react at room temperature with 125 ml of plant extract and 500 ml of silver nitrate solution was added to 1000 ml of flask and allowed to react under room conditions. AgNPs formation was monitored primarily with a macroscopic method depending on the change of colour and was determined according to time with spectrophotometric measurements (Pugazhendhi et al., 2018). Composed dark colored solution was obtained with centrifuge (7500 rpm, 10 min). The top, liquid portion was removed, while the remaining solid segment was washed with distilled water. The resulting AgNPs were left to dry at 65 °C and stored in the dark until use in the characterization process.

Characterisation techniques

Synthesised silver nanoparticles (AgNP) UV-Vis. Spectrums were obtained on a spectrophotometer (UV-1601 220 V SHIMADZU). Scanning electron microscope (EVO 40 LEQ) was used to measure AgNPs dimension and morphology of AgNPs crystal construction was analysed with X-ray diffractometer (RadB-DMAX II) between $3^{\circ} \le 2\theta \ge 80^{\circ}$ spaces. SEM- energy dispersive X-ray (EDX) spectroscopy was used to demonstrate AgNPs existence and morphology. FTIR (Perkin Elmer ATR-FTIR) analysis was used to determine those functional groups that has a role in the reduction in the plant extract in the range of 4000-400 cm⁻¹. Thermogravimetric analysis (TGA SHIMADZU DTG-60H) reported the degradation temperatures of AgNPs under the atmosphere of nitrogen gas at 25 °C min⁻¹ and the atmosphere of nitrogen gas at constant heating rate.

Antimicrobial activity of AgNPs

The effects of the obtained particles on gram negative Escherichia coli ATCC 25922, gram positive Staphylococcus aureus ATCC 29213 bacteria and Candida albicans yeast were investigated. Minimum Inhibitory Concentration (MIC) was determined by microdilution method. In practice, the muller Hilton medium was added to the microplate wells and incubated overnight at 37 °C by addition of an appropriate amount of the microorganism mixture and the AgNP solution adjusted to 0.5 in turbidity according to the Mc Farland standard (El-Batal et al., 2018; Vishwasrao et al., 2018; Dhand et al., 2016). The next day the MIC was determined by the well in which the reproduction ended. In addition, vancomycin, colistin and fluconazole commercial

antibiotics and 1 mM AgNO₃ solution were used for *S. aureus*, *E. coli* and *C. albicans* to compare the effects of AgNPs.

Result and discussion

XRD analysis with X-ray diffraction

The evaluation of the analysis of AgNPs XRD phase and crystal structure was shown in *Figure 1*, which was synthesised with green method. Crystal structures of silver nanoparticles' reflections (111), (200), (220) and (311) were calculated with the values 2θ (37.38°, 44.78°, 64.81° and 77.50°) on XRD analysis (*Fig. 1*). It was determined that AgNPs were of elemental (Ag°) and cubic crystalline structure. The crystalline particle size of the AgNPs was calculated according to the Debye-Scherrer equation and found to be approximately 27.30 nm.

The Debye-Scherrer equation is (Narayanan and Sakthivel, 2011; Banala et al., 2015; Rolim et al., 2019):

$$D = K\lambda / (\beta \cos\theta)$$

Here, D is the crystal size of AgNPs (nm), k = Scherrer's constant, (0.90), $\lambda =$ X-ray wavelength (1.5406° A), $\beta =$ FWHM (full width at half maximum) of the peak located at $2\theta = 37.38^\circ$, $\theta =$ Bragg's angle of diffraction.

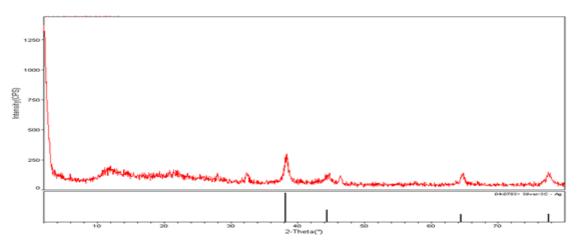


Figure 1. XRD measurements of AgNPs confirming the cubic crystalline structure of AgNPs

Analysis of UV-vis spectroscopy

In UV-vis spectroscopy analysis silver nanoparticles' formation was observed with samples taken at different times at 0, 5, 10, 15, 20, 25, 30 and 45 min. The color change to dark brown, which shows the formation of AgNPs, was observed very quickly. At about a maximum of 441.58 nm a sharp plazmon resonance indicated appropriately synthesised AgNPs (*Fig. 2*) Similarly, some other researchers which were near to our research found different spectrum values at 390-500 nm surface plazmon resonance (Al-Bahrani et al., 2017; Kalimuthu et al., 2008; Anbu et al., 2018; Khan et al., 2018; Rajakumar et al., 2017).

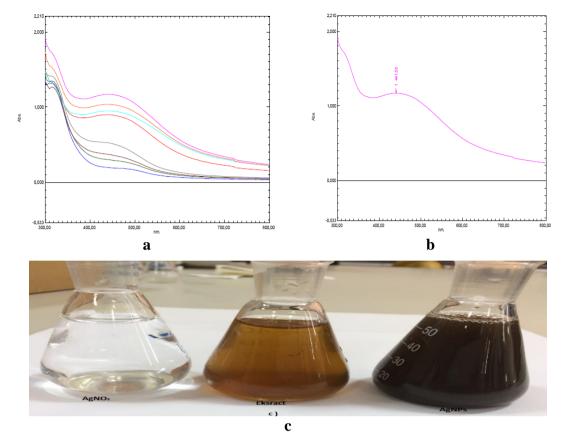


Figure 2. a The UV–Vis spectra of AgNPs synthesized by Cydonia oblonga leaf aqueous extract. b On UV-Vis spectroscopy maximum absorbance value synthesised AgNPs. c AgNPs stepwise thematic representation of the synthesis carried out

Fourier transform infrared spectroscopy (FTIR) spectroscopy analysis

FT-IR spectrum provided information on which functional groups the reaction occurs in by comparing *Cydonia oblonga* leaf extract and synthesised AgNPs (*Fig. 3*). The characteristic peaks appeared at 3291 cm⁻¹ and it represents the -OH stretching vibrations of free and hydrogen bound hydroxyl groups present in *Cydonia oblonga* extract. The peak obtained at 2081 cm⁻¹ is due to -CH stretching vibrations of methylene groups, and furthermore, the sharp peak present at 16365 cm⁻¹ can be assigned to non-hydrated C=O groups and it revealed that the asymmetric and symmetric stretching vibrations of carboxyl groups.

FTIR spectrum measurements which were synthesised on different studies showed the existence of intense band on 3314, 2119 and 1636 cm⁻¹ (Pugazhendhi et al., 2018; Baran et al., 2018).

Analysis results of silver nanoparticles SEM and EDAX

Morphological specifications of AgNPs' which are obtained from leaf extract of *Cydonia oblonga* were examined by using electron microscope (SEM), shown in *Figure* 4. The results have shown the presence of silver nanoparticles (AgNPs). Morphological specifications which were obtained from leaf extract of *Cydonia oblonga* were examined by using electron microscope (SEM), shown in *Figure* 4. The results have

shown us clearly the existence of spherical shaped silver nanoparticles (AgNPs') under 100 nm. It was seen on XRD calculation that particles size was about 27.30 nm. Similarly, SEM analysis provided supplementary information about AgNPs size and morphological specifications (De Jesús Ruíz-Baltazar et al., 2017; Prakash et al., 2013; Jha and Shimpi, 2018).

Energy distributed spectrums of acquired AgNPs from EDX analysis have shown that there have been pure silver pieces (*Figure 5*). The results obtained with AgNPs in other scientific studies support our findings (Veisi et al., 2018; Kumar et al., 2016; Ramkumar et al., 2017).

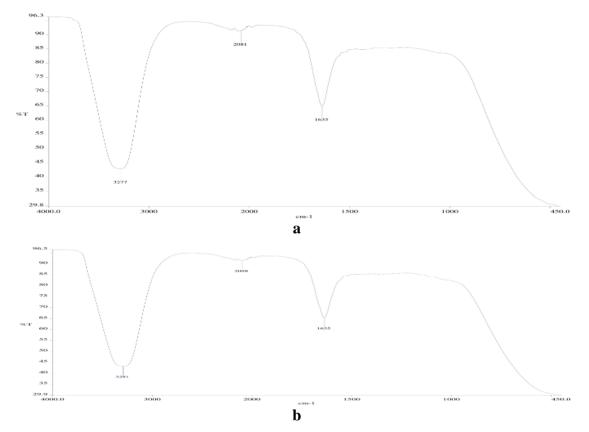
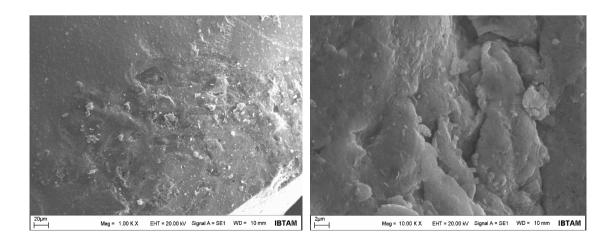


Figure 3. a FT-IR spectrum of plant extract spectrum. b FT-IR spectrum of synthesised AgNPs



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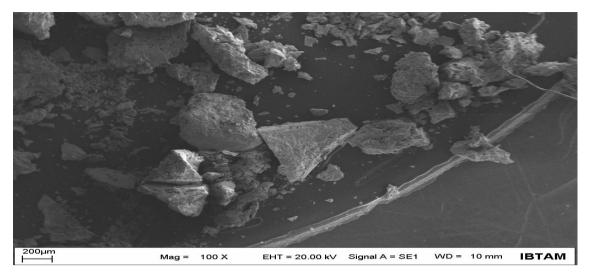


Figure 4. Images of synthesised silver nanoparticles of SEM analyses

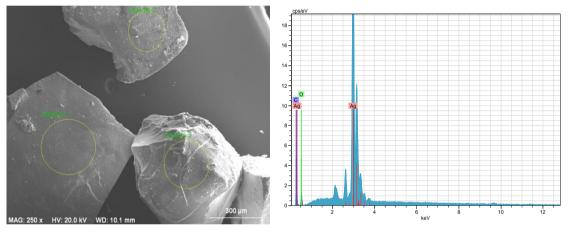


Figure 5. EDX profile of AgNPs showing its elemental composition

TGA-DTA findings

Nanoparticles which were prepared with green synthesis were analysed with the flow rate of 20 mL min⁻¹ in N₂ (g) atmosphere with 10 °C min⁻¹ heating rate between 31-900 °C TGA and DTA data. TGA bent shows specimen mass loss for thermal degradations. DTA bent determines maximum heat of dissociation at every stage of degradation (Baran et al., 2018).

It has been seen in *Figure 6* that at 31-257 °C mass loss is derived from moisture, at 257-333 °C there is a mass loss which is derived from cellulosic materials, at 333-900 °C the mass loss is derived from plant extract phytochemicals. TGA data were reported on the study about AgNPs synthesised mass loss. Similar studies support the results in this aspect (Baran et al., 2018).

Antimicrobial activity of synthesised AgNPs

Microorganisms are more and more resistant to the antibiotics used, which causes the struggle with them to fail. Therefore, the search for antimicrobial agent is more

interested in combating infection. Antimicrobial activity studies with AgNPs may be a serious alternative to this condition. AgNPs increase the formation of reactive oxygen species (ROS) and disrupt the wall structure. These (ROS) to the affinity of the membrane structure and function changes are produced. They have an antimicrobial effect because they form a repressive process (Wang et al., 2017).

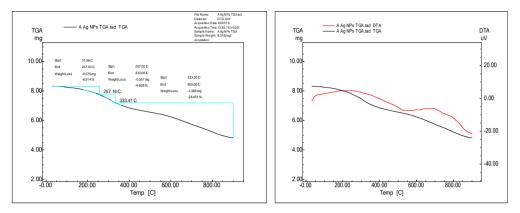


Figure 6. TGA-DTA analysis result of synthesised silver nanoparticles

In our study, the effect of AgNPs' antimicrobial effect is searched which are obtained from leaf extract of *Cydonia oblonga* by MIC method. Antifungal activities of AgNPs were studied on *C. Albicans* and probable antibacterial effect on gram positive *S. aureus ATCC 29213* and gram negative *E. coli ATCC 25922*. MIC worths were determined of AgNP and 1 mM silver nitrate. In our study, fluconazol, vancomisin and Colistin antibiotics were used. The synthesized silver nanoparticles have been shown to have a strong antimicrobial effect. The MIC of the AgNPs on *E. coli ATCC 25922*, *S. aureus ATCC 29213* and *C. albicans* was determined as 0.0552, 0.1535 and 0.0383 mg mL⁻¹, respectively. From the acquired results, which have been shown in *Table 1*, it is clear that AgNPs are more effective than antibiotics and 1 mM silver nitrate solution.

In order to determine the antimicrobial activity of AgNPs, in the other scientific studies, 53, 15 and 40 μ g mL⁻¹ MIC values were found for *E. Coli* (Sinsinwar et al., 2018; Rolim et al., 2019; Hemmati et al., 2019). Values of 100 and 11.60 μ g mL⁻¹ were calculated for E. coli (Ananda et al., 2019; Shao et al., 2018).

Table 1. MIC values of synthesized silver nanoparticles (AgNPs) (mg mL⁻¹), silver nitrate and S. aureus, C. Albicans, vacomycin, fluconazole, colistin antibiotics for E. coli, respectively

Organism	AgNPs	Silver nitrat	Antibiotic
S. aureus ATCC 29213	0.1535	0.500	0.50
C. albicans	0.0385	0.500	0.50
E. coli ATCC25922	0.0552	1.000	0.125

Conclusions

As a result, silver nanoparticles were characterized by UV-visible spectra, TGA-DTA, FTIR, SEM-EDX and XRD measurements. The nanoparticles were obtained by eco-friendly synthesis method with the use of *Cydonia oblonga* leaf aqueous extract, which does not involve simple, easy and toxic chemicals. In the case of waste, the use of the leaves as a vegetable source for synthesis provides a great advantage both for the recycling of the waste and for the cheap and easy production of the raw material for synthesis. AgNPs obtained by environmentally friendly methods can act as a powerful antimicrobial agent and their contribution to human health in medical industry can be a serious factor. The utilization of the leaf in waste form in this way and the synthesis will be quite interesting considering the transfer of the recycling to the industry. These nanoparticles can also be used for dye removal and bioremediation of waste water.

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