CHARACTERIZATION AND EVALUATION OF HEAVY METAL POLLUTION IN SOIL-WHEAT SYSTEM AROUND COAL MINES IN PINGDINGSHAN, CHINA

 $\label{eq:2.1} \begin{array}{l} Z\text{Hang, W. P.}^{1}-\text{Qian, J.}^{1}-\text{Xu, G. J.}^{1}-\text{Zhang, D. M.}^{1}-\text{Kang, C.}^{1}-\text{Feng, D. X.}^{1}-\text{Shi, L.}^{1}-\text{Zhang, C. L.}^{1,2^{*}}-\text{Guo, Z. Y.}^{1^{*}}-\text{Ma, J. H.}^{1,2}-\text{Zhang, C. S.}^{1} \end{array}$

¹College of Environment and Planning, Henan University, Kaifeng 475001, China

²Institute of Natural Resources and Environment, Henan University, Kaifeng 475001, China

³School of Geography and Archaeology, National University of Ireland, Galway, Ireland

*Corresponding author

e-mail: zhangcl@henu.edu.cn, guozy9888@163.com; phone: +86-371-2388-1850; fax: +86-378-2388-1850

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Abstract. Indoor experiments were conducted to fully understand soil metal contamination around coal mines in Pingdingshan, Henan province, China. Forty-three paired soil and wheat product samples were collected from three main mining areas in Pingdingshan. Soil Cd, Zn and Pb were assessed using the diffusive gradients in thin films (DGT) and pentetic acid or diethylene triamine penlaacetic acid (DTPA). The pollution levels were evaluated. The Nemero comprehensive pollution index shows the soil heavy metal pollution is slight; the single coefficient of potential ecological hazard indicates the soil pollution degree of heavy metals ranks as Cd > Pb > Zn, which is the most serious pollution. Cd is close to medium ecological hazard, and other heavy metals are minor ecological hazards. **Keywords:** *coal mines, wheat, DGT, DTPA, potential ecological risk*

Introduction

Mineral resources are the basic source of human production and life, and an important material basis for socioeconomic development. As the demand for energy has increased since the industrial revolution, the human society has increasingly exploited fossil energy such as coal. The development of mineral resources plays an important role in promoting the national economy growth, but has also brought some environmental pollution problems. During the mining process, heavy metals in coal and coal gangue can enter and the soil around mining areas through leaching, infiltration, migration and other ways. Heavy metal pollution in the soil affects the soil microbial activity and thus the structure and function of the soil ecosystem. Heavy metals in the soil enter the food chain through food and vegetables, causing health risks to humans.

Various heavy metal environmental pollutants have received much attention in recent years because of their persistent bioaccumulation and high toxicity.

Some heavy metals such as Zn and Cu above the safe concentration can be toxic, while other heavy metals can be harmful even at low concentrations. As reported, the farmland soil around a coal mine in north Bangladesh was severely contaminated by heavy metals such as Mn, Zn, As, Pb and Ti (Bhuiyan et al., 2010). Behaddya et al. used GIS to characterize the spatial distribution of heavy metal pollution in the surface soil of a coal mine area in Algeria (Behaddya and Hadjel, 2014). However, the effective assessment of health risks to residents is of very insufficient. In this research, The

objective of this study is used diffusive gradients in thin-films (DGT) to assess the heavy metal contents around the mining areas in Pingdingshan and to evaluate the heavy metal pollution levels and environmental risks.

Materials and methods

Sampling and sample pretreatment

Pingdingshan (33°08′-34°20′ N, 112°14′-113°45′ E) in Henan Province is known as the coal bunker of middle China. It is also an important base of the energy and raw material industry. In this study, 42 paired samples of soil and wheat were collected from mining areas of Pingdingshan in June 2017. The sampling sites were chosen near three local mines (Hugou-RA, Zaoyuan-RB, Bishan mines-RC). The sample number and GPS coordinates of Pingdingshan area are shown in *Table 1*. The soil samples were air-dried and then passed through < 2 mm, < 0.25 mm and < 0.15 mm sieves. The wheat samples were washed with deionized water, separated into roots and straws, and dried at 75 °C until constant weight.

Sample name	GPS coordinates	Sample name	GPS coordinates
RA-1	33.7576929013,113.3456307421	RB-7	33.7772575173,113.3553975642
RA-2	33.754610741,113.3564701998	RB-8	33.7720293931,113.3683628691
RA-3	33.7498132802,113.3494517421	RB-9	33.757254742,113.3572521998
RA-4	33.754610741,113.3564701998	RB-10	33.7669667372,113.3614036575
RA-5	33.7498132801,113.3494517422	RB-11	33.755858741,113.3606531998
RA-6	33.7498132802,113.3494517421	RB-12	33.7672397372,113.3615656575
RA-7	33.7528111219,113.3536091998	RB-13	33.7772575173,113.3553975642
RA-8	33.7525591219,113.3533481998	RB-14	33.757254741,113.3572521998
RA-9	43.2086443635,116.6611639959	RB-15	33.7772575173,113.3553975642
RA-10	33.767433359,113.3566805642	RC-01	33.7501588052,113.3748027507
RA-11	33.7621259603,113.3745851151	RC-02	33.7503669635,113.3684036575
RA-12	33.7672397372,113.3615656575	RC-03	33.7503669635,113.3684036575
RA-13	33.7672397372,113.3615656575	RC-04	33.7580789603,113.3724291151
RA-14	33.7731903931,113.3681138691	RC-05	33.7580789603,113.3724291151
RA-15	33.7772575173,113.3553975642	RC-06	33.7580789603,113.3724291151
RB-1	33.757254741,113.3572521998	RC-07	33.7503669635,113.3684036575
RB-2	33.7660787372,113.3618146575	RC-08	33.755858741,113.3606531998
RB-3	33.757254741,113.3572521998	RC-09	33.7549999603,113.3749691151
RB-4	33.757254741,113.3572521998	RC-10	33.7550169603,113.3749571151
RB-5	33.766244359,113.3566705642	RC-11	33.7549549603,113.374636112
RB-6	33.757254741,113.3572521998	RC-12	33.7551279603,113.3750131151

Table 1. Sample number and GPS coordinates in Pingdingshan area

Soil physicochemical analyses

Soil pH was measured by a pH detector in a 1:2.5 soil/water suspension (w/v). SOM (soil organic matter) was determined using the potassium dichromate volumetric method. Particle size was analyzed using a Mastersizer 3000 device (Cai et al., 2015).

Soil available Cd

Since it is available Cd rather than total Cd that matters in Cd toxicity, here available Cd was focused. Cd extracted by passive sampling methods such as diethylene triamine penlaacetic acid (DTPA) and DGT can be considered as available Cd in this work.

DGT deployment

DGT developed in 1994 by British scientists Davison and Zhang Hao is a contaminant in-situ research technique based on Fick's first law of diffusion. With DGT, the bioavailability of heavy metals can be studied by simulating the heavy metal absorption by plants or other organisms, and the results reflect both static processes (soil particles and soil solutions) and dynamic processes (Luo et al., 2011).

Compared with other traditional morphological analysis techniques, DGT can more efficiently measure the biological effective states of heavy metals in the nature and better reflect the heavy metals absorbed by organisms.

The DGT device consists of an innermost adsorption layer, a diffusion layer covered with specific thickness, and protective filter membranes (Wei et al., 2018).

As the adsorbent in the adsorption film has strong binding capacity with specific phosphorus and metals (metalloid), the target substance (analyte) in the solution can be rapidly and irreversibly fixed on the binding phase and consumed through the diffusion membrane.

Based on Fick's first law (Zhou et al., 2011), the diffusion flux F of the analyte is:

$$\mathbf{F} = \mathbf{D}\partial\mathbf{C} / \partial\mathbf{x} \tag{Eq.1}$$

where D is the diffusion coefficient of the tested substance in the diffusion membrane; C is the concentration of the test substance in the solution.

The time-averaged concentration of the target substance measured by DGT can be calculated as follows:

$$C_{DGT} = M\Delta g / DAt$$
 (Eq.2)

where M is the amount of the target substance in the adsorbent, Δg is the thickness of the diffusive layer, A is the area of the DGT device window, and t is the adsorption time.

Soil and heavy metal determination

Air-dried soil samples over a 2-mm nylon sieve were weighed (20 g) and placed into 250-mL plastic containers (Fan et al., 2010).

Then deionized water with the maximum field water holding capacity of 70% was added under full stirring. The plastic wrap was covered to prevent moisture and evaporation, followed by equilibration at 25 °C constantly for 48 h.

After that, a small amount of the treated soil (about 3 g) was placed with a clean plastic spoon into the round hole of the DGT device and gently shaken parallel on the table surface, making the soil fully contact with the filter surface. Soil was further added until the inner cavity was filled.

The loaded DGT device was transferred to a ziplock bag added with a small amount of deionized water in advance, and the bag mouth was semi-closed (Xu et al., 2004).

After placement at a constant temperature for 24 h, the soil was removed and the DGT device was rinsed with deionized water. Then the fixed membrane was removed into the ziplock bag, and minor deionized water was dripped to wet it. After sealing, it was stored at 4 °C for analysis.

Single and comprehensive pollution indices

The soil pollution degree of each metal was measured using the single pollution index (P_i), which is the ratio of the heavy metal concentration in the contaminated soil to that in the reference soil. P_i was calculated as follows (Cai et al., 2015; Yang et al., 2012):

$$P_i = \frac{c_i}{c_o} \tag{Eq.3}$$

where C_i and C_o are the heavy metal concentrations in the contaminated soil and the reference soil, respectively. The soil comprehensive pollution status for all the heavy metals was calculated as the Nemero comprehensive pollution index P_c . The result not only considers the average pollution level of various heavy metals, but also indicates the most serious pollution in the soil contributed by single heavy metals. P_c was calculated as follows (Zhou et al., 2011; Chen et al., 2014; Gil-Sotres et al., 2004):

$$P_c = \sqrt{\frac{(\overline{P_i})^2 + \max(P_i)^2}{2}}$$
(Eq.4)

Potential ecological hazard index

The potential ecological hazard index is based on the characteristics, environmental migration and sedimentation of heavy metals and evaluates heavy metals in soil or sediments from the perspective of sedimentology. This index considers the soil contents of heavy metals and combines ecology, environmental science and biotoxicology (Fan et al., 2003; Li et al., 2014). It comprehensively analyzes the heavy metal migration and transformation in soils and sediments and evaluates the sensitivity of geographical regions to heavy metals and the differences in background values of geographical regions. The degree of potential ecological hazard eliminates the influence of geographical differences (Xiong et al., 2017; Wang et al., 2017), reflects the relative contribution of biological effectiveness and spatial differences, and uncovers the potential impact of heavy metals on the ecoenvironment. It is suitable for largescale comparison of sediments and soils (Xu et al., 2004). Relevant formulas are:

$$RI = \sum_{i=1}^{n} E_r^i$$
 (Eq.5)

$$E_r^i = T_r^i \times C_f^i \tag{Eq.6}$$

$$C_f^i = \frac{c_s^i}{c_n^i} \tag{Eq.7}$$

APPLIED ECOLOGY AND ENVIRONMENTAL RESEARCH 17(3):5435-5447. http://www.aloki.hu • ISSN 1589 1623 (Print) • ISSN 1785 0037 (Online) DOI: http://dx.doi.org/10.15666/aeer/1703_54355447 © 2019, ALÖKI Kft., Budapest, Hungary where E_r^i is a potential ecological hazard individual coefficient; T_r^i is the toxicity response coefficient of a certain metal based on Hakanson's standardized heavy metal toxicity coefficient (Cd = 30, Pb = 5, Zn = 1); C_r^i is the single pollution coefficient; C_s^i is the measured heavy metal content in surface soil; C_n^i is the reference value (from National Secondary Standard for Soil Environmental Quality).

Results and discussion

Soil properties and contents of metals

The general characteristics of the soils are shown in *Table 2*. Soil pH ranges from 5.2 to 8.3. In RA, the soils are weakly alkaline, with pH between 7.5 and 8.3. In RB, the majority of the soils are acidic, with pH between 5.2 and 6.6. As for RC, the pH is mainly in the neutral range, with the mean of 7.3. The soil organic carbon (OC) in the research areas ranges between 1.9% and 9.2%, which indicates the difference in soil fertilization degree. The Cd content ranges between 0.13 and 0.4 mg·kg⁻¹, with only one sample from RB exceeding farmland soil pollution risk screening value (GB 15618-2008). The Zn content ranges between 50.7 and 125.1 mg·kg⁻¹, while the Pb content ranges between 15.5 and 38.2 mg·kg⁻¹.

Sitor			$\mathbf{OC}(0)$	Total o		oncentration (mg·kg ⁻¹)			
Siles		рп	UC (%)	(Cd	Zn		Pb	
D۸	Mean	7.8	5.1	0	.16	69.8			34.8
KA	Range	7.5 - 8.3	2.6 - 9.2	0.15	- 0.19	61.9 - 7	9.8	30	.8 - 38.2
DD	Mean	5.8	4.1	0	.18	67.9			22.6
Range		5.2 - 6.6	1.9 - 6.6	0.13	3 - 0.4	50.7 - 125.1		15	.5 - 35.5
DC	Mean	7.3	5.4	0	.15	66.9			26.6
KC	Range	6.7 - 8.3	2.9 - 7.3	0.1	- 0.3	57.4 - 9	4.9	17	.7 - 35.7
	Soil back	ground value of He	enan Province ¹			0.064	62.	.5	21.8
						0.25 ^a	150) a	80 ^a
Standard limit	nit Environ	montal quality stan	dard for $soils^2$			0.30 ^b	200) ^b	80 ^b
	Environ	mental quanty stand				0.45 °	250) c	80 °
						0.80^{d}	300) d	80 ^d

 Table 2. General characteristics of soils collected from the study area

¹The soil background value of Henan province is derived from literature (Shao et al., 1998)

²The standard value of soil environmental quality originates from the secondary standard of soil environmental quality standard (GB15618-2008), a is soil pH < 5.50, b is pH between 5.50 and 6.50, c is soil pH between 6.50 and 7.50, and d is the corresponding quality standard when soil pH > 7.50

Table 3 shows the metal contents measured by DTPA and DGT. The ranges of Cd, Zn and Pb are 2.5-11.1, 85.3-520.2 and 0.65-4.88 μ g·kg⁻¹ respectively (DTPA), and 2.3-5.5, 57.1-491.3 and 1.1-9.4 μ g·kg⁻¹ respectively (DGT). DTPA has higher extraction capacity than DGT, which may be related to the different chemical properties of the extractants. DTPA is an efficient chelating agent with strong extraction ability and can extract water-soluble, exchange, mineral-bound and organic complex heavy metals in soil.

Sites			CDTPA (µg·kg ⁻¹)	Cdgt (µg·kg ⁻¹)			
Siles		Cd	Zn	Pb	Cd	Zn	Pb	
DA	Mean	4.0	190.0	3.8	3.3	176.3	4.0	
KA	Range	3.5 - 6.1	85.3 - 300.1	2.7 - 4.9	2.3 - 5.5	88.8 - 491.3	1.9 - 8.2	
חח	Mean	5.1	171.1	3.04	2.5	76.5	2.5	
KD	Range	3.1-11.1	61.7 - 412.3	0.7 - 3.8	3.5 - 5.5	76.5 - 242.0	1.9 - 5.50	
DC	Mean	5.0	337	3.4	3.0	115.7	4.5	
ĸĊ	Range	2.5 - 6.8	41.5 - 520.2	0.8 - 4.5	2.4 - 4.3	57.1 - 178.3	1.1 - 9.4	

Table 3. Contents of metals measured using DTPA (C_{DTPA}) and DGT technique (C_{DGT})

Tables 4 and 5 show the bioavailable concentrations of Zn, Cd, Pb in the DGT resin eluent and the concentrations of heavy metals measured by DTPA. DTPA extractable heavy metals fraction, as a form of heavy metals with high biological effectiveness, can be used as an effective index of heavy metal pollution in similar soil by its rapid and convenient leaching (Yang et al., 2012; Lu and Zhao, 2017). DTPA can extract watersoluble, exchangeable, adsorptive, organic stationary and partially-oxidized heavy metals from soil, which is considered to be a highly bioactive form related to its total amount. The available heavy metals measured by DGT include the free state of soil solution and the unstable organic and inorganic complex state released from the surface of soil particles (Chen et al., 2012; Bernalte et al., 2013). The average extraction rates (i.e. the percentage of extracted heavy metal content in the total amount of this heavy metal) of Zn, Cd and Pb are 3.18%, 30.46% and 13.03%, respectively in DTPA, and 0.21%, 2.01% and 0.02%, respectively in DGT.

C:4ag	CDGT			Сдтра			
Siles	Zn	Cd	Pb	Zn	Cd	Pb	
1	117.25	2.27	8.25	2.39	0.04	4.00	
2	126.16	3.68	3.38	2.49	0.04	4.88	
3	491.39	3.60	2.09	1.70	0.04	3.02	
4	316.63	5.52	2.89	3.01	0.04	4.65	
5	153.30	2.87	4.98	2.11	0.04	4.49	
6	120.56	2.94	2.39	1.25	0.04	2.93	
7	123.72	2.82	2.68	1.63	0.04	4.09	
8	163.01	3.71	5.13	2.14	0.04	4.47	
9	116.85	2.79	1.94	2.52	0.04	3.84	
10	161.59	2.50	3.97	0.97	0.05	2.99	
11	136.34	2.79	4.97	0.85	0.05	2.67	
12	88.76	4.44	5.48	1.70	0.06	3.48	
13	136.10	3.22	3.85	1.39	0.05	3.19	
14	97.44	2.75	4.49	0.61	0.03	0.65	
15	177.21	3.28	5.49	1.55	0.04	2.30	
16	132.24	2.48	4.95	1.29	0.05	2.85	
17	184.63	12.01	5.03	1.69	0.07	3.10	
18	162.93	2.65	3.62	1.09	0.05	2.30	
19	124.35	2.82	3.59	0.89	0.05	3.18	
20	101.86	2.85	3.86	0.98	0.05	3.77	

Table 4. C_{DGT} and C_{DTPA} of Zn, Cd and Pb

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21	187.55	4.11	4.45	1.56	0.03	3.39
22	139.18	3.15	5.24	1.41	0.05	3.58
23	130.74	2.63	5.13	1.40	0.06	3.79
24	76.51	3.25	4.97	4.37	0.05	2.59
25	128.21	2.72	2.49	1.48	0.06	3.10
26	153.62	3.17	5.54	1.82	0.05	3.37
27	98.63	3.02	2.55	4.72	0.11	3.54
28	159.61	5.57	3.53	1.23	0.05	3.52
29	76.29	3.54	2.54	2.40	0.05	3.53
30	116.06	2.44	9.41	3.67	0.05	3.51
31	57.08	2.52	1.15	5.20	0.06	3.66
32	116.30	2.80	5.36	2.37	0.05	3.43
33	140.60	2.56	3.97	2.15	0.05	3.34
34	150.31	2.45	7.24	1.71	0.04	2.63
35	143.91	2.74	5.81	2.79	0.05	3.09
36	166.08	3.75	4.42	0.41	0.02	0.83
37	108.88	4.34	3.52	4.74	0.06	3.67
38	70.18	3.31	2.26	3.84	0.06	3.99
39	96.73	2.77	4.12	4.84	0.07	3.93
40	178.31	2.73	6.15	4.69	0.05	4.50

Continue table 4. C_{DGT} and C_{DTPA} of Zn, Cd and Pb

Statistic	CDGT			Сдтра			
Staustic	Zn	Cd	Pb	Zn	Cd	Pb	
Maximum	491.39	12.01	9.41	5.20	0.11	4.88	
Minimum	57.08	2.27	1.15	0.41	0.02	0.65	
Average	143.18	3.39	4.32	2.23	0.05	3.35	
Standard deviation	70.41	1.57	1.67	1.29	0.01	0.85	
Coefficient of variation	0.49	0.46	0.39	0.58	0.29	0.25	

Table 6 presents the soil particle concentration (P_c) in each sampling point, the Rdiff calculated using the 2D DIFS model, and the effective concentration (CE) calculated as the ratio between C_{DGT} and Rdiff. In general, soil P_c in the same area is approximately the same, with the lowest SA. The CE ranks as Zn > Pb > Cd, while the heavy metal content extracted by DTPA ranks as Pb > Zn > Cd.

The relationships between Cd, Zn, Pb in the roots, straws and that extracted by DGT and DTPA are shown in *Figure 1* and *Table 7*. Although the C_{DGT} of Cd, Zn and Pb in the soil is one to two orders-of-magnitude lower than those extracted by DTPA, C_{DGT} extraction is closer to wheat availability than DTPA extraction, and DTPA extraction may overestimate the concentration of available states, and some of the extracted metals may not be absorbed by wheat. More studies also show DGT better reflects plant uptake than chemical extraction. For example, PUEYO M used DGT to assess the influence factors on cadmium accumulation in rice grains from paddy soils of three parent materials and found that cadmium bioavailability could well be assessed by DGT (Pueyo et al., 2008). Bernd Nowack et al. used DGT to test undisturbed field soils and suggested that DGT can be successfully used under undisturbed field conditions to study the kinetics of metal resupply.

Second a stars			R _{diff}		$C_E(\mu \mathbf{g} \cdot \mathbf{L}^{-1})$			
Sample sites	$P_c \mathbf{g} \cdot \mathbf{cm}^{-3}$	Cd	Zn	Pb	Cd	Zn	Pb	
1	20.9	0.28	14.86	1.37	2.2652	117.2454	8.2474	
2	17.4	0.455	15.99	0.562	3.68095	126.1611	3.38324	
3	12.9	0.445	62.28	0.348	3.60005	491.3892	2.09496	
4	23.4	0.682	40.13	0.48	5.51738	316.6257	2.8896	
5	18.2	0.355	19.43	0.828	2.87195	153.3027	4.98456	
6	24.9	0.363	15.28	0.397	2.93667	120.5592	2.38994	
7	13.3	0.349	15.68	0.445	2.82341	123.7152	2.6789	
8	26.1	0.459	20.66	0.852	3.71331	163.0074	5.12904	
9	19.4	0.345	14.81	0.322	2.79105	116.8509	1.93844	
10	16	0.309	20.48	0.659	2.49981	161.5872	3.96718	
11	20.4	0.345	17.28	0.826	2.79105	136.3392	4.97252	
12	15.63	0.549	11.25	0.91	4.44141	88.7625	5.4782	
13	22.9	0.398	17.25	0.639	3.21982	136.1025	3.84678	
14	28.3	0.34	12.35	0.746	2.7506	97.4415	4.49092	
15	29.2	0.406	22.46	0.912	3.28454	177.2094	5.49024	
16	32.1	0.306	16.76	0.822	2.47554	132.2364	4.94844	
17	20.1	1.484	23.4	0.836	12.00556	184.626	5.03272	
18	20.4	0.328	20.65	0.602	2.65352	162.9285	3.62404	
19	35.7	0.348	15.76	0.596	2.81532	124.3464	3.58792	
20	31.8	0.352	12.91	0.641	2.84768	101.8599	3.85882	
21	28.2	0.508	23.77	0.739	4.10972	187.5453	4.44878	
22	21.4	0.389	17.64	0.87	3.14701	139.1796	5.2374	
23	22.2	0.325	16.57	0.852	2.62925	130.7373	5.12904	
24	22.65	0.402	9.697	0.826	3.25218	76.50933	4.97252	
25	22	0.336	16.25	0.414	2.71824	128.2125	2.49228	
26	27.7	0.409	30.71	0.878	3.30881	242.3019	5.28556	
27	13.2	0.373	12.5	0.423	3.01757	98.625	2.54646	
28	24.2	0.688	20.23	0.587	5.56592	159.6147	3.53374	
29	25.8	0.437	9.669	0.422	3.53533	76.28841	2.54044	
30	24.6	0.301	14.71	1.563	2.43509	116.0619	9.40926	
31	23.3	0.311	7.234	0.191	2.51599	57.07626	1.14982	
32	21.6	0.346	14.74	0.89	2.79914	116.2986	5.3578	
33	24	0.316	17.82	0.66	2.55644	140.5998	3.9732	
34	25.7	78.033	872	1.202	631.287	6880.08	7.23604	
35	26.4	0.339	18.24	0.965	2.74251	143.9136	5.8093	
36	24.5	0.464	21.05	0.735	3.75376	166.0845	4.4247	
37	18.7	0.536	13.8	0.584	4.33624	108.882	3.51568	
38	27	0.409	8.895	0.376	3.30881	70.18155	2.26352	
39	33.8	0.342	12.26	0.685	2.76678	96.7314	4.1237	
40	21.7	0.337	22.6	1.022	2.72633	178.314	6.15244	

Table 6. R_{diff} and C_E of metals

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Figure 1. Relationship between heavy metals extracted by DGT and DTPA: relationship between the contents of (a) Zn, (b) Cd, (c) Pb in roots and from DGT and DTPA; relationship between the contents of (d) Zn, (e) Cd and (f) Pb in straw and from DGT and DTPA

Environment evaluation

Single and comprehensive pollution indices

The results of heavy metal pollution index are shown in *Table 8*. In terms of Zn, more than half of the samples are at light pollution, while others are still at clean level. In terms of Cd, most of the samples are in mild pollution (Liu et al., 2017; Shi et al., 2013), and only 9.76% of them are in moderate pollution. Noticeably, 3 samples are in severe pollution, which indicates high risks. As for Pb, 53.69 of the samples are in moderate pollution index of 43 samples ranges from 1.22 to 3.36 with a mean of 1.7 < 2, indicating the soil is at slight pollution, and Cd is under the severest pollution.

The E_r^i and RI were calculated by *Equations 3, 4* and 5 (*Table 9*). The single coefficient of potential ecological hazard and other elements belong to minor ecological hazards. Chu Chunjie et al. pointed out that Cu, Zn, Ni and Pb induced chronic non-

carcinogenic health risks (Chen et al., 2017; Bhuiyan et al., 2010), while Cr had a clear risk of cancer in the soil of hilly slopes around mining areas in Pingdingshan. The results indicate Zn and Pb in the soil around the mining areas of Pingdingshan exceed the standard, which is a non-carcinogenic health risk (Yenilmez et al., 2011).

Table 7. Relationship between heavy metals extracted by different extractants and that in wheat

		Linear equation	r	р
	Zn _{DGT}	y = 0.6694 x + 114.59	0.19	< 0.01
	Zn _{DTPA}	y = 0.0013 x + 1.9466	0.01	< 0.01
Dest	Cd_{DGT}	y = 2.9972 x + 2.6501	0.34	< 0.01
Root	Cd _{DTPA}	y = 0.0109 x + 0.0455	0.12	< 0.01
	Pb_{DGT}	y = 0.371 x + 2.801	0.32	< 0.01
	Pb _{DTPA}	y = 0.0231 x + 1.9169	0.03	< 0.01
	Zn _{DGT}	y = 0.1744 x + 92.806	0.33	< 0.01
	Zn _{DTPA}	y = 0.0585 x + 1.1555	0.27	< 0.01
Ctrown	Cd_{DGT}	y = 3.7437 x + 2.3191	0.51	< 0.01
Straw	Cd _{DTPA}	y = 0.1624 x + 0.0259	0.49	< 0.01
	Pb _{DGT}	y = 0.3985 x + 2.8309	0.61	< 0.01
	Pb _{DTPA}	y = 0.0187 x + 3.0632	0.19	< 0.01

Table 8. Heavy metal pollution index evaluation results

Heavy metal element	Zn	Cd	Pb
Clean (a)	0	0	0
Clean (%)	0	0	0
Still clean (a)	15	0	0
Still clean (%)	36.59	0	0
Mild pollution (a)	26	34	19
Mild pollution (%)	63.41	82.93	46.34
Moderate pollution (a)	0	4	22
Moderate pollution (%)	0	9.76	53.69
Severe pollution (a)	0	3	0
Severe pollution (%)	0	7.32	0

Table 9. Farmland soil E^{i}_{r} , RI in Pingdingshan mining area

		Ni		7-	Dh	
E ^{Cu} r	$E^{\rm Cr}r$	$E^{m}r$	$E^{Cu}r$	$E^{Z_{n}}r$	Erbr	RI
37.8	0.5	0.5	1.6	0.3	4.5	45.2

General situation of heavy metal pollution in soil

The results showed that the average pH values of wheat fields in the study area were 7.8, 5.8 and 7.3, respectively, in which RA and RC were at the alkaline level and RB was at the acidic level. According to the statistical results in *Table 2*, the average content of heavy metal elements in the wheat fields in the three mining areas all

exceeded the background value of soil elements in Henan province. The average values of Cd, Zn and Pb in the wheat fields in RA mining area were 2.50, 2.81 and 2.34 times of the background value, respectively. The average values of Cd, Zn and Pb in the wheat field in the RB mining area were 1.12, 1.09 and 1.07 times of the background values, respectively. The average values of Cd, Zn and Pb in wheat field in RC mining area were 1.60, 1.04 and 1.22 times of the background values, respectively. According to the grade ii standard of soil environmental quality standard (GB15618-2008), the average values of Cd, Zn and Pb in wheat field in the RB mining area were 0.20, 0.23 and 0.44 times of the standard values, respectively. The average values of Cd, Zn and Pb in the wheat field in the RB mining area were 0.60, 0.34 and 0.28 times of the standard values, respectively. RC mining area of wheat to Cd, Zn and Pb respectively, the average standard value of 0.33, 0.27, and 0.33 times. The soil heavy metal pollution degree of RB > RC > RA, and by a single heavy metal pollution degree, Cd > Pb > Zn, along with the exploitation of coal mine, the content of soil heavy metal levels in gradually rise, which is consistent with the research results of Chen et al. (2017).

Conclusions

In this paper, three typical coal mines in Pingdingshan were investigated to study the heavy metal pollution in the surrounding soil and wheat using the diffusive gradients in thin films (DGT) and pentetic acid or diethylene triamine penlaacetic acid (DTPA), and to explore its potential ecological risks. The results show that the soil heavy metal pollution around the main mining areas of Pingdingshan is under different levels. The average soil contents of Zn, Cd and Pb are 68.64, 0.68 and 26.96 mg·L⁻¹, respectively. According to the extraction effect of different extractants. DTPA has higher extraction capacity than DGT, which may be related to the different chemical properties of the extractants. From the average extraction efficiency of different extraction methods for the same heavy metal. The average extraction rates (i.e. the percentage of extracted heavy metal content in the total amount of this heavy metal) of Zn, Cd and Pb are 3.18%, 30.46% and 13.03%, respectively in DTPA, and 0.21%, 2.01% and 0.02%, respectively in DGT. The Nemero comprehensive pollution index shows the soil pollution of heavy metals is slight. The single ecological coefficient of potential ecological hazard indicates the soil pollution degrees of various heavy metals rank as Cd > Pb > Zn, and the most serious pollution is Cd, which is close to medium ecohazards, while other heavy metals are under minor ecological hazards.

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